



Module (Course Syllabus) Catalogue 2022-2023

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|--------------------------|--|-------------------------|
| College/ Institute | Erbil Medical Technical Institute | |
| Department | MLT Department | |
| Module Name | Clinical Chemistry 2 | |
| Module Code | CLC401 | |
| Degree | Technical Diploma <input checked="" type="checkbox"/> Bachler <input type="checkbox"/> High Diploma <input type="checkbox"/> Master <input type="checkbox"/> PhD <input type="checkbox"/> | |
| Semester | 4rd | |
| Qualification | Master degree | |
| Scientific Title | lecturer | |
| ECTS (Credits) | 5 | |
| Module type | Prerequisite <input type="checkbox"/> Core <input checked="" type="checkbox"/> Assist. <input type="checkbox"/> | |
| Weekly hours | 4 | |
| Weekly hours (Theory) | (2)hr Class | (3)Total hrs Workload |
| Weekly hours (Practical) | (2)hr Class | (1)Total hrs Workload |
| Number of Weeks | 16 | |
| Lecturer (Theory) | Muharam Yaseen Mohammed | |
| E-Mail & Mobile NO. | Muharam.mohammed@epu.edu.iq (07504490568) | |
| Lecturer (Practical) | | |
| E-Mail & Mobile NO. | | |
| Websites | https://academicstaff.epu.edu.iq/faculty/muharam.mohammed | |

Course Book

| | | | | | |
|------------------------------------|--|----------------|-----------------------|-----------------|----------------------------------|
| Course Description | This course aims to provide comprehensive theoretical knowledge in clinical chemistry including kidney, liver and limits for all chemical tests, diagnosis and disease treatment and advanced practical training in this diverse field. | | | | |
| Course objectives | Upon completion of the course students will 1. Have advanced knowledge on the methodology of clinical chemistry. 2. Be able to understand the synthesis of laboratory kits. 3. Have advanced skills in blood treatment, analysis and disease diagnosis. | | | | |
| Student's obligation | 1. The student attention in all theoretical and practical lectures in academic year. 2- Completion of all tests. 3- Attendance in exams 4. Write or prepare reports. | | | | |
| Required Learning Materials | Data show, lab. For practical part | | | | |
| Evaluation | | Task | Weight (Marks) | Due Week | Relevant Learning Outcome |
| | | Paper Review | 1 | 1 | |
| | Assignments | Homework | 0.5 | 4 | |
| | | Class Activity | 2 | 2 | |
| | | Report | 1 | 1 | |
| | | Seminar | 1 | 1 | |
| | | Essay | 0 | 0 | |
| | | Project | 0 | 0 | |
| | | Quiz | 1 | 4 | |
| | | Lab. | 2 | 12 | |
| | | Midterm Exam | 1 | 2 | |
| | | Final Exam | 1 | 3 | |
| | | Total | | | |

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|---|--|-------------------------|
| Specific learning outcome: | <p>On successful completion of this program, graduates will be able to:</p> <p>Identify, evaluate and apply major theoretical traditions in clinical chemistry studies.</p> <p>Understand how the human body work.</p> | |
| Course References: | <p>Text book of Clinical Chemistry and Molecular Diagnostics Seventh Edition Carl A . Burits , David E .Bruns</p> | |
| Course topics (Theory) | Week | Learning Outcome |
| Body fluid and electrolyte, Intracellular fluid compartment | 1 | |
| Extracellular fluid compartment, Electrolytes | 2 | |
| Electrolytes are in inside and outside of the cell | 3 | |
| Methods of distribution of the electrolytes inside and outside of the cells | 4 | |
| Electrolyte function of Sodium | 5 | |
| Electrolyte function of Potassium | 6 | |
| Electrolyte function of Bicarbonate CO ₂ | 7 | |
| Electrolyte function of Chloride | 8 | |
| The Liver | 9 | |
| liver function tests | 10 | |
| Bilirubin | 11 | |
| Direct and indirect bilirubin | 12 | |

| Practical Topics | Week | Learning Outcome |
|---|------|------------------|
| Creatinine Test | 1 | |
| Blood Urea Nitrogen (BUN) | 2 | |
| examination | 3 | |
| Electrolyte test (in chemical methods) | 4 | |
| Electrolyte test (in instrument) | 5 | |
| liver function tests :Alanine transaminase (ALT) test or(GPT) | 6 | |
| Aspartate aminotransferase (AST) test or(GOT) | 7 | |
| Alkaline phosphatase (ALP) test | 8 | |
| Albumin test | 9 | |
| Bilirubin test | 10 | |
| examination | 11 | |
| Review | 12 | |

Questions Example Design

rite the differences of the following.

| B.urea | Creatinin |
|--------|-----------|
| | |

| Osmtic pressure | Diffusion property |
|-----------------|--------------------|
| | |

Q2 / Define the following.

1-GOT

2-Alkaline phosphates

3. Filtration:

4-Hypertonic solutions:

5- Hypernatremia:

Q3/ Fill the blanks of the following sentences by suitable words.

1-The blood sugar level is the amount of ----- in the blood, and normally levels stay limits morning to -----mg/dl .

2- Extracellular fluid contain ----- and ----- ions, while intracellular fluid contain ----- and ----- ions.

3-ALT is used by the body to metabolize ----- and its normal value in the blood is -----

4-Diffusion methods depending to transferring molecules and ions from----- to----- .

Extra notes:

External Evaluator

The outcome of course book evaluation is commonly more explicit and follows the principles and rules in general.

**Sangar sabah sabir
Lecturer**